SYNTHESIS OF NEW SUBSTITUTED 3-MERCAPTO-1,2,4-TRIAZOLES POSSESSING 5-H-DIBENZO|a,d|CYCLOHEPTENE MOIETIES.

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ABSTRACT

Seven new thiosemicarbazides 2a-g bearing 5H-dibenzo[a,d]cycloheptenyl-methyl group at N¹ and different alkyl or aryl substituents at N⁴ were synthesized using classical procedures. ¹H-NMR analysis indicated the existence of two conformational isomers, a major axial (about 75%) and a minor equatorial one (25%) which are interconvertible by middle ring inversion. Cyclization of 2a-g in NaOH solution afforded the corresponding 3-mercapto-1,2,4-triazoles 3a-g which were separated as pure axial isomers. All the new compounds were extensively characterized by IR-, UV-, ¹H-NMR and ¹³C-NMR spectroscopy and will be screened as carbonic anhydrase inhibitors in order to test the application of "bulky scaffold" strategy in this class of compounds.

INTRODUCTION

Continuing our investigations in the domain of biologically active compounds¹⁻⁴ we have shown recently⁵ that aryl sulfonylureido-glycine hydroxamates containing a bulky 5H- dibenzo[a,d]cycloheptene moiety indicated a strong inhibition towards matrix metalloproteinases (MMP) and <u>Clostridium histolyticum</u> collagenase. In this preliminary paper we describe the synthesis and characterization of new 3-mercapto-1,2,4-triazoles containing the above-mentioned 5H-dibenzo[a,d]cycloheptene moiety. The new compounds will be biologically tested as carbonic anhydrase inhibitors (CAI) in order to see if the "bulky scaffold" strategy applied to MMP⁵ could be expanded to CAI also.

RESULTS

The synthesis of mercaptotriazoles 3a-g was performed using the reaction sequence presented in Scheme 1.

The known ester $1b^6$ of the 5H-dibenzo[a,d]cycloheptenyl-acetic acid $1a^7$ was converted to the corresponding hydrazide 1c (new compound, mp=161-163 0 C; NMR δ values 8.7 and 9.3 ppm for NH and NH₂ protons; UV(MeOH+2%DMSO): 239nm and 291nm) on treatment with hydrazine hydrate in boiling ethanol (20hrs). Reaction of 1c with alkyl- and aryl-isothiocyanates in anhydrous ethanol at reflux (6hrs) afforded the new thiosemicarbazides 2a-g with yields varying between 76% and 97%. Treatment of 2a-g with excess

2N-NaOH solution at reflux (6hrs) produced the cyclization to the novel 3-mercapto-1,2,4-triazoles 3a-g which were precipitated with acetic acid in yields around 70%).

The structures of thiosemicarbazides 2 were confirmed by their spectral data. Thus, in the IR spectra (FTS-135 BioRad apparatus; KBr) characteristic absorptions appear at 3110-3350 cm⁻¹ (three bands; ν_{NH}); 1670-1690 cm⁻¹ (ν_{CO}); along with all the frequencies characteristic for hydrocarbon moieties. All the compounds 2a-g afford very similar UV spectra (MeOH+2%DMSO) exhibiting two major absorptions at 240-247 nm and 278-290.5nm. The ¹H-NMR spectra (VARIAN GEMINI 300MHz apparatus, d₆-DMSO) indicated the presence of two isomers, 2-axial and 2-equatorial in about 3:1 ratio (interconvertible by middle ring inversion).

In 2-axial isomers the H^{5'} (eq)* is deshielded, appearing as a triplet at 4.62-4.65ppm (J = 7.0-7.3Hz), whereas the CH₂^{12'} protons shielded by the double bond show a doublet at 2.57-2.60ppm. A reversal of this situation occurs at 2-equatorial isomers where the H^{5'} (ax) appears as a shielded triplet at 3.73-3.76 ppm (J = 7.8-8.1Hz) and the CH₂^{12'} protons as a doublet at 3.40-3.41ppm. The NH protons signals appear as singlets between 8.90-10.25ppm, the double bond protons H^{10'} and H^{11'} as singlets at 7.03-7.21 whereas the aromatic protons occur in the usual interval (7.10-7.60ppm). The aliphatic side chains protons indicated also normal δ values. In the ¹³C-NMR spectrum of 2a-g (VARIAN GEMINI apparatus, 75MHz, d₆-DMSO) the dibenzo[a,d]cycloheptene moiety appears in a narrow δ domain (126-131ppm) with C^{10'} and C^{11'} easily recognizable at δ = 130.8ppm (A signal at δ ~181ppm could be attributed to C=S carbon atom). The remaining carbon atoms afford signals which partially overlap with those of the side chains. However the attributions of these signals could be made using incremental calculations for substituents and the results are very similar to those of mercaptotriazoles 3a-g (see Table 2).

* For numbering of the skeleton see Tables 1 and 2

H	'	'	•	1.84	2.41	2.47	3.89
Н 20,			•	7.18-7.46 m	7.10-7.40 m	7.10-7.31 m	6.96 d (8.9)
Hil		,		7.18-7.46 m	7.10-7.40 m	6.79 d (8.1)	6.771
AH.	•		0.90t (7.3)	7.18-7.46 m	7.10-7.40 m		
Htř		4.22 Id(17.3) 5.16Id(10.3)	1.26 m	7.18-7.46 ш		6.79 d (8.1)	6.77 d (8.9)
H16'	1.09 t (7.1)	5.70 ddt (17.3, 10.8 5.3)	1.45 ш		7.10-7.40m	7.10-7.31 m	6 96 d b 96 9
1, J Hz)	3.70 q (7.1)	4.29ld (5.31)	3.59 t (7.7)		,	,	٠
ISO, δ ppπ H ¹²	3.09 d (7.7)	3.08 d (7.7)	3.08 d (7.7)	2.89 dd (5.2, 7.80) 2.78 dd (5.2, 7.8)	2.97 d (7.9)	2.97 d (7.9)	2.94 d (7.8)
3a-g (d ₆ -DM H ¹⁰ H ¹¹	7.01 s	7.00 s	7.02 s	6.67 d(12.0) 6.60 d(12.0)	6.62 s	6.66 s	6.67 s
triazoles H ⁵	4.61 t (7.7)	4.58t (7.7)	4,57 t (7.7)	4.511 (7.8)	4.381 (7.9)	4.37 t (7.9)	4.411
or mercapto H ¹⁻² H ⁶⁻³	7.20-7.40 m	7.20-7.38 m	7.15-7.36 m	7.18-7.46 m	7.10-7.40 m	7.10-7.31 m	7.15-7.28 m
H-NMR datz	IS IS CH2-CH3	LS 16 17 CH ₂ -CH=CH ₂	U IF IF IF CH2-CH3-CH3	H ₃ C ₁ C ₁ C ₁ C ₁ C ₂ C ₂ C ₃ C ₄	15 CH ₃	15 15 15 15 15 15 15 15 15 15 15 15 15 1	15 (17) 21.
Table 1	3a	36	3c	34	3e	3f	38
	1 ¹ H-NMR data for mercaptotriazoles 3a-g (d ₆ -DMSO, δ ppm, J Hz) R H ¹⁻⁴ H ³ H ⁴³ H ⁴³ H ⁴³ H ⁴⁵ H	¹ H-NMR data for mercaptotriazoles 3a-g (d ₆ -DMSO, δ ppm, J Hz) R H ¹⁻⁴ H ¹⁵ H ¹⁶ H ¹⁷ H ¹⁷ H ¹⁸	H-NMR data for mercaptotriazoles 3a-g (d ₆ -DMSO, δ ppm, J Hz) H ¹⁵ H ¹⁶ H	H-NMR data for mercaptotriazoles 3a-g (d ₆ -DMSO, δ ppm, J Hz) Hi ¹⁶ Hi ¹⁶	H-NMR data for mercaptortriazoles 3a-g (d ₆ -DMSO, δ ppm, J Hz)	H-NMR data for mercaptotriazoles 3a-g (d ₂ -DMSO ₁ , S ppm, J Hz)	H-NMR data for mercaptorriazoles 3a-g (d _e -DMSO, \hat{O} ppm, J H \hat{D}) H ¹⁰ H

10.80

10.72

11.02

10.80

10.80

138.4

138.5

								in A	io I	GE CH2		±s-						
4	ble 213 C-NMR data for mercaptotriazoles 3 (de-DMSO, δ ppm, J Hz)	or mer	captot	riazoles	3 (de-D)	δ ,08M	ppm, J	Hz)										
_	R	C _s ,	C ¹²	C11.	သ	ည	່ເງັນ	ر _ي دور	ر _ي ري	ر ر	CIS	Cle	Cl"	C ^{I8}	C19,	C ^{20′}	$C^{2l'}$	C _{II} O
	15' 16' CH ₂ -CH ₃	52.3	26.1	131.0	150.8	166.1	130.0	129.2	127.3	129.8	38.6	13.6						133.
	15' 16' 17' CH ₂ -CH=CH ₂	52.2	26.4	130.9	151.4	1.791	130.0	129.2	127.2	129.7	45.4	130.4	118.2					133.
	15. 16. 17. 18. CH2-CH2-CH3	52.2	26.2	131.0	151.2	6.991	130.0	128.3	127.2	129.5	43.5	30.2	19.9	13.6		,		133.
	H ₃ C ₁₅ 17	52.2	25.8	130.6	151.4	167.4	130.0	128.6	127.0	129.0	138.6	131.8	129.7	129.6	127.2	127.2	17.4	133.
	ST S	52.8	26.0	130.6	151.5	6.791	129.9	129.0	127.0	129.5	139.6	128.5	132.9	129.7	129.4	125.0	21.13	133.
	15 21. 20 19	52.5	26.0	130.6	151.5	168.0	130.2	129.4	127.1	129.9	140.0	127.0	127.7	130.4	127.7	127.0	21.4	133.
	15 21. 27. 28. 38. 30. 31.	52.4	25.8	130.5	151.0	8.791	129.7	128.7	126.7	129.1	138.5	129.3	114.5	160.0	114.5	129.3	55.4	133.

138.6

138.6

138.6

138.5

The structures of the mercaptotriazoles 3 were also proved on the basis of their spectral data. The IR spectra proved the cyclization of 2 to 3 by the disappearance of ν_{CO} band. Moreover instead of three NH bands of 2, compounds 3 show (in KBr) only two bands at 3110-3130 and 3380-3450 cm⁻¹ characteristic for -NH-CS. It seems that in KBr the mercaptotriazoles 3 exists in the form of their thionic tautomers:

In the UV spectra (MeOH+2%DMSO) the triazoles 3a-g indicated only two important absorption maxima: compounds 3a-c at 255-256nm and 291-292nm and aromatic substituted compounds 3d-g at 230-242nm and 262-267nm with a shoulder at about 287nm. The ¹H-NMR spectra of triazoles 3a-g (Table 1) indicated the presence of a single conformational isomer, namely the axial one (corresponding to 2-axial). The H⁵ (eq) appears at δ =4.38-4.60ppm (triplet, J = 7.7Hz) and CH₂¹² protons as a doublet at 3.10ppm in 3a-c and at 2.80-2.95ppm in 3d-g. Interestingly, in d₆-DMSO solution, the NH signals of 2a-g totally disappeared being replaced by a singlet at δ =10.8-11.8ppm attributable to SH proton. Thus, in solution the above tautomeric equilibrium is shifted towards the thiolic form. A similar shift of the thione-thiole tautomerism was observed by us at mercaptotriazoles bearing phenazinic moieties⁸. Conversion of thiosemicarbazides 2 to mercaptotriazoles 3 is nicely supported also by ¹³C-NMR spectra (Table 2) where a new quaternary carbon signal (for C³) appears at δ =166-168ppm simultaneously with the disappearance of the C=S signal from 2 (δ =181ppm). Moreover, instead of C=O signal from 2 at 169-170ppm a new signal appears for C⁵ of 3 at about 151ppm. The complete ¹H and ¹³C-NMR data for 3a-g are given in Tables 1,2.

DISCUSSION

Occurrence of thiosemicarbazides 2 in two conformational isomers, 2-axial, 2-equatorial, is not unexpected keeping in mind some previously reported results⁹ when a similar slow conformational equilibrium was proved for 5-methyl-5H-dibenzo[a,d]cycloheptene and studied by variable temperature NMR.

Cyclization of thiosemicarbazides to mercaptotriazoles in alkaline medium could be explained as in other related cases³ by the mechanism outlined in Scheme 2 having as essential intermediate the tautomeric anion 5.

Scheme 2

To our surprise cyclization of 2 to 3 and the subsequent work-up were accompanied by the loss of minor equatorial isomer probably more soluble in acidic water. The resulting solid mercaptotriazoles 3 are pure axial isomers.

In conclusion, in this paper we described the synthesis and spectral characterization of seven new thiosemicarbazides and seven new mercaptotriazoles possesing the bulky 5H-dibenzo[a,d]cycloheptene moiety. The occurrence of thiosemicarbazides 2a-g in two conformational isomeric forms was proved and a thiole-thione tautomeric equilibrium dependent on the physical state was observed for triazoles 3. Compounds 3 will be biologically screened as CAI in order to see if the "bulky scaffold" strategy is applicable in this domain.

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